Interaction of the Components in the System Ba-Tb-Cu-O and Related Systems

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The interaction of the components in the A-R-Cu-O systems, where A is an alkaline-earth and R is a rare-earth metal, has attracted considerable interest for a long time. The compounds of these systems are usually characterized by layered structures and are first of all interesting as high-temperature superconductors. The most prominent among them is undoubtedly YBa₂Cu₃O₇ (*YBCO*), the crystal structure and properties of which have been studied meticulously. The interaction of the components in the system Ba-Y-Cu-O and related systems with other alkalineearth and rare-earth metals has also been widely investigated [1]. In this work we focus on the system Ba-Tb-Cu-O, and some related systems, for which data in the literature are scarce.

Polycrystalline samples of the BaO-Tb₂O₃-CuO system were prepared by solid-state reaction of appropriate amounts of high-purity carbonates and oxides at 1173 K in air in two 24-hour stages. Firstly the reagents were mixed and heated in a corundum crucible to decompose the carbonates. Then, after cooling to room temperature, the mixtures were reground to achieve homogeneity, pressed into pellets and sintered again. Phase and structure analyses were carried out on X-ray powder diffraction data (DRON 2.0M diffractometer, Fe K_{α} radiation). Samples of nominal composition $A_{0.5}$ Tb_{0.5}TO₃, where A =Ca, Sr or Ba, T = Ni or Zn, were prepared in similar way.

The isothermal section of the phase diagram of the BaO-Tb₂O₃-CuO system at 1173 K was constructed based on the investigation of fifteen samples. The existence of the compounds BaTbO₃ (SrZrO₃ structure type, *I4/mcm*, a = 0.6034(1), c = 0.8576(1) nm, $R_B = 0.052$) and Ba₄₄Cu₄₅O₉₀ (own structure type, *Im-3m*, a = 1.8266(3) nm, $R_B = 0.237$) was confirmed in the pseudo-binary BaO-Tb₂O₃ and BaO-CuO systems, respectively. It was established that the interaction of the components in the ternary region of the BaO-Tb₂O₃-CuO system is characterized by the absence of compounds. The only compound reported in the literature, TbBa₂Cu₃O₇ with orthorhombic YBa₂Cu₃O₇ structure type [2], was not observed under the conditions of our study. The BaTbO₃ phase is in equilibrium with the Ba₄₄Cu₄₅O₉₀ and CuO phases. Hence, there are three three-phase and two two-phase regions in the system. It should be noted that, during the synthesis, the initial cubic Tb₂O₃ phase with (Mn_{0.5}Fe_{0.5})₂O₃ structure type was oxidized into the rhombohedral Tb₇O₁₂ phase with Pr₇O₁₂ structure type.

Ba_{0.5}Tb_{0.5}ZnO₃ and Sr_{0.5}Tb_{0.5}ZnO₃ were found to be two-phase samples, which contained in equilibrium BaTbO₃ (SrZrO₃ structure type, *I4/mcm*, a = 0.6039(1), c = 0.8578(1) nm, $R_B =$ 0.099), or SrTbO₃ (GdFeO₃ structure type, *Pnma* a = 0.5945(1), b = 0.8347(2), c = 0.5879(1)nm, $R_B = 0.160$), and ZnO. The investigation of the samples with nominal compositions Ca_{0.5}Tb_{0.5}ZnO₃ and Ca_{0.5}Tb_{0.5}NiO₃ showed the presence of phases that correspond to the reagents only – CaO, Tb₇O₁₂ and ZnO (or NiO), in appropriate ratios.

^[1] P. Villars, K. Cenzual (Eds.), Pearson's Crystal Data – Crystal Structure Database for Inorganic Compounds, ASM International, Materials Park, OH, USA, Release 2014/15.

^[2] H. Ouchi, Preparation and superconducting properties of *Ln*-Ba-Cu-O ceramics, *Ferroelectrics* **95** (1989) 215-220.